

Peng-Wu Zheng,^{a*} Wei Wang^b
and Xue-Min Duan^a^aSchool of Pharmacy, Jiangxi Science and
Technology Normal University, Nanchang
330013, People's Republic of China, and^bDepartment of Chemical Engineering, Anshan
University of Science and Technology, Anshan
114002, People's Republic of China

Correspondence e-mail: zhengpw@sohu.com

Key indicators

Single-crystal X-ray study

 $T = 294$ KMean $\sigma(\text{C}-\text{C}) = 0.003$ Å R factor = 0.038 wR factor = 0.116

Data-to-parameter ratio = 11.9

For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.*N,N'*-Bis(3-nitrobenzylidene)hydrazine

The title compound, $\text{C}_{14}\text{H}_{10}\text{N}_4\text{O}_4$, was synthesized by the reaction of 3-nitrobenzaldehyde with hydrazine hydrate. The molecule possesses a crystallographically imposed center of symmetry and is essentially planar; the dihedral angle between the nitro group and the benzene ring is $3.4(2)^\circ$. Intermolecular $\text{C}-\text{H}\cdots\text{O}$ hydrogen bonds link the molecules into an infinite chain.

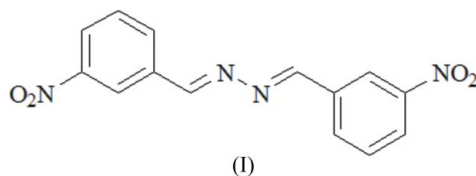
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Comment

Recently, a number of azine compounds containing both a diimine linkage and N–N bonding have been investigated in terms of their crystallography and coordination chemistry (Zheng *et al.*, 2005; Kundu *et al.*, 2005; Kesslen & Euler, 1999; Armstrong *et al.*, 1998; Xu *et al.*, 1997). We report here the crystal structure of the title compound, (I), where two 3-nitrobenzylidene units are directly linked through the imine N atoms.



The title molecule crystallizes in an *E,E* configuration, possessing a crystallographically imposed center of symmetry at the mid-point of the N–N bond (Fig. 1). This configuration agrees with that commonly found in a number of azine compounds (Zheng *et al.*, 2005; Şengül *et al.*, 2004; Liu *et al.*, 2004). The N–N bond length of 1.409 (3) Å (Table 1) is somewhat longer than that observed in related azine compounds (Xu *et al.*, 2005; Şengül *et al.*, 2004). The $\text{C}=\text{N}-\text{N}$ angle [$111.9(2)^\circ$] is similar to that in *N,N'*-bis(4-chloro-

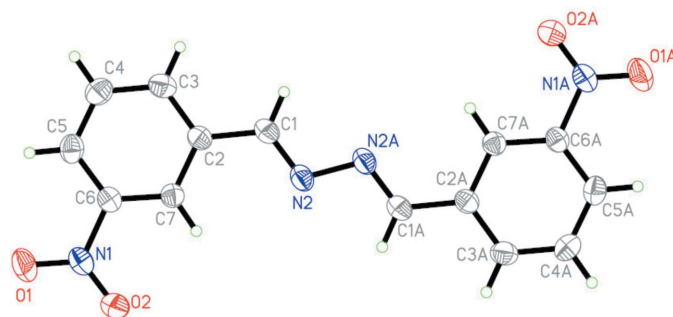


Figure 1

View of the molecule of (I), with displacement ellipsoids drawn at the 30% probability level. The suffix A corresponds to symmetry code $(3-x, 1-y, 2-z)$.

benzylidene)hydrazine (Zheng *et al.*, 2005), but significantly smaller than the ideal sp^2 value of 120° , as a consequence of repulsion between the nitrogen lone pairs and the adjacent $C=N$ bond. The dihedral angle between the nitro group and the benzene ring is $3.4(2)^\circ$. The bis-benzylidenehydrazine backbone assumes a planar structure, with an r.m.s. deviation of $0.012(2) \text{ \AA}$.

In the crystal structure, a weak intermolecular $C-H \cdots O$ hydrogen bond (Table 2) leads to a ten-membered ring, described by the graph-set descriptor $R_2^2(10)$, linking the molecules into an infinite chain (Fig. 2).

Experimental

The title compound was synthesized by the reaction of 3-nitrobenzaldehyde with hydrazine hydrate in refluxing ethanol (Liu *et al.*, 2004). Single crystals suitable for X-ray analysis were obtained by slow evaporation of a tetrahydrofuran solution.

Crystal data

$C_{14}H_{10}N_4O_4$	$D_x = 1.447 \text{ Mg m}^{-3}$
$M_r = 298.26$	Mo $K\alpha$ radiation
Monoclinic, $P2_1/n$	Cell parameters from 1134 reflections
$a = 6.9694(19) \text{ \AA}$	$\theta = 3.0\text{--}26.1^\circ$
$b = 7.833(2) \text{ \AA}$	$\mu = 0.11 \text{ mm}^{-1}$
$c = 12.990(4) \text{ \AA}$	$T = 294(2) \text{ K}$
$\beta = 105.175(4)^\circ$	Block, light yellow
$V = 684.4(3) \text{ \AA}^3$	$0.34 \times 0.30 \times 0.14 \text{ mm}$
$Z = 2$	

Data collection

Bruker SMART 1000 CCD area-detector diffractometer	1204 independent reflections
φ and ω scans	814 reflections with $I > 2\sigma(I)$
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)	$R_{\text{int}} = 0.035$
$T_{\text{min}} = 0.962$, $T_{\text{max}} = 1.000$	$\theta_{\text{max}} = 25.0^\circ$
3318 measured reflections	$h = -8 \rightarrow 8$
	$k = -7 \rightarrow 9$
	$l = -15 \rightarrow 13$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.061P)^2 + 0.0625P]$
$R[F^2 > 2\sigma(F^2)] = 0.038$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.116$	$(\Delta/\sigma)_{\text{max}} < 0.001$
$S = 1.03$	$\Delta\rho_{\text{max}} = 0.12 \text{ e \AA}^{-3}$
1204 reflections	$\Delta\rho_{\text{min}} = -0.18 \text{ e \AA}^{-3}$
101 parameters	Extinction correction: SHELXL97
H-atom parameters constrained	Extinction coefficient: $0.094(9)$

Table 1

Selected geometric parameters (\AA , $^\circ$).

O1—N1	1.223(2)	N2—C1	1.272(2)
O2—N1	1.2181(19)	N2—N2 ⁱ	1.409(3)
N1—C6	1.473(2)		
O2—N1—O1	123.11(17)	O1—N1—C6	118.20(17)
O2—N1—C6	118.68(16)	C1—N2—N2 ⁱ	111.9(2)

Symmetry code: (i) $-x + 3, -y + 1, -z + 2$.

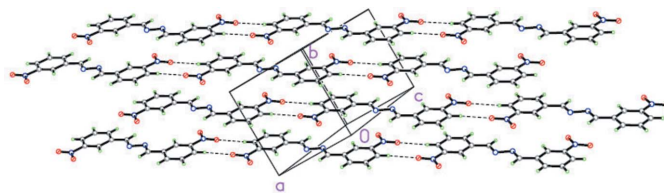


Figure 2

The packing of (I), showing the intermolecular hydrogen-bonded (dashed lines) extended network.

Table 2

Hydrogen-bond geometry (\AA , $^\circ$).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
$C5-H5 \cdots O1^{ii}$	0.93	2.56	3.487(3)	176

Symmetry code: (ii) $-x + 1, -y + 2, -z + 2$.

All H atoms were positioned geometrically and refined as riding, with $C-H = 0.93 \text{ \AA}$ and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$.

Data collection: SMART (Bruker, 1997); cell refinement: SAINT (Bruker, 1997); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Bruker, 1997); software used to prepare material for publication: SHELXTL.

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